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### Nature Of Solids Section Review

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### Nature Of Solids Section Review Answers

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### Nature Of Solids Section Review Answers

review key concepts in Section 13.3. 12 3 L2 L1 Section 13.3 Assessment 15. Particles in solids are packed tightly together in an orderly arrangement. The locations of the particles are fixed. 16. Section 13.3 The Nature of Solids 399 Non-Crystalline Solids Not all solids are crystalline in form: some solids

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solids are dense and not easy to compress. Because the particles in solids tend to vibrate about fixed points, solids do not flow. When you heat a solid, its particles vibrate more rapidly as their kinetic energy increases. The organization of particles within the solid breaks down, and eventually the solid melts.

### 133. The Nature of Solids

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### Chapter 13 - States of Matter - 13.3 The Nature of Solids ...

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### 13.3 The Nature of Solids Flashcards | Quizlet

Solids, liquids and gases The particle theory is used to explain the properties of solids, liquids and gases. The strength of bonds (attractive forces) between particles is different in all three ...

### Solids - Solids, liquids and gases - KS3 Chemistry ...

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Solids. In the solid state, the individual particles of a substance are in fixed positions with respect to each other because there is not enough thermal energy to overcome the intermolecular interactions between the particles. As a result, solids have a definite shape and volume. Most solids are hard, but some (like waxes) are relatively soft.

### 12.2. Properties of Liquids and Solids - Chemistry LibreTexts

SECTION 13.3 THE NATURE OF SOLIDS (pages 396-399) This section describes the highly organized structures of solids, distinguishes between a crystal lattice and a unit cell, and explains how allotropes of an element differ. A Model for Solids (page 396) 1. Is the following sentence true or false? Although particles in solids have kinetic ...

### Name Date Class STATES OF MATTER 13

Section 2.1 Liquids, Solids, and Gases Goals To describe a model that allows you to visualize the particle nature of matter. To describe the similarities and differences among solids, liquids, and gases in terms of this model. This is a very important section because it presents a model that you will use throughout your

### Chapter 2 The Structure of Matter and the Chemical Elements

Section 3.5 Common Elements Goal: To apply the information described in the first four sections of this chapter to the description of some common elements. This section brings the chapter full circle back to the particle nature of solids, liquids, and gases, but after reading this section, you will know more about the particles that compose

### Chapter 3 The Structure of Matter and the Chemical Elements

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13.3- The Nature of Liquids 13.4 Phase Digrams Solids- definite volume and shape, high density Solids and Liquids have high densities because their molecules are close together. Solids Intermolecular forces are strong Molecules still move! Can only vibrate and revolve in place.

### Chapter 13- The States of Matter 13.1- The Nature of Gases

This page contains a collection of printable materials for teaching students about states of matter (solid, liquid, gas). Includes a cut-and sort activity, higher-level thinking questions, and more.