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**Nuclear physics - Wikipedia**

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This article summarizes equations in the theory of nuclear physics and particle physics. Definitions. Quantity (common name/s) (Common) symbol/s Defining equation SI units Dimension Number of atoms  $N =$  Number of atoms remaining at time  $t$ .  $N_0 =$  Initial number of atoms at time  $t = 0$   $N_D =$  Number ...



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A is alpha particle, B is neutron, C is proton and D is electron. Question 2. A radon specimen emits radiation of  $3.7 \times$

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10<sup>3</sup> GBq per second.

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Nucleus consists of:  $Z$  protons with  $e^+$  charge  
 $N$  neutrons with no charge.  $A$

Mass number  $A=Z+N$  protons &

neutrons are bound by strong force:  $R$

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10<sup>-13</sup> m. Nomenclature and common units  
Power Prefix Abbrev. 10<sup>-15</sup> femto f  
10<sup>-12</sup> pico p 10<sup>-9</sup> nano n 10<sup>-6</sup> micro m  
10<sup>-3</sup> milli m 10<sup>-2</sup> centi c 10<sup>-1</sup> deci d 10<sup>3</sup>  
kilo k

### **1.3. Basic Principles of Nuclear Physics**

[<sup>14</sup>N(n,p) C, the neutrons being

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produced by cosmic rays].  $^{14}\text{C}$  decays by  $\beta$ -emission with half life  $T_{1/2} = 5730$  y  $^{14}\text{C} \rightarrow ^{14}\text{N} + \beta^- + \bar{\nu}$  The chemical activity of  $^{14}\text{C}$  is similar to that of  $^{12}\text{C}$ , so living organisms have the same  $^{14}\text{C}/^{12}\text{C}$  ratio as in the atmosphere, which is about  $1.35 \times 10^{-12}$ . When an organism dies it stops absorbing  $^{14}\text{C}$ , and the ratio

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binding energy/ nucleon of Ne 20, He 4  
and C 12 are 8.03 MeV, 7.07 MeV and  
7.86 MeV respectively.

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