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as suggest alternative methods to solve particular problems to the teachers. The target is to direct individuals towards problem solving strategies, rather than solving problems in one prescribed format.

NCERT Solutions for Class 12 Chemistry (Updated for 2020-21)

Chemistry Notes for class 12 Chapter 2 Solutions Solution is a homogeneous mixture of two or more substances in same or different physical phases. The substances forming the solution are called components of the solution. On the basis of number of components a solution of two components is called binary solution. Solute and Solvent

Chemistry Notes for class 12 Chapter 2 Solutions

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Chemistry Revision Notes for Class 12 Chapter 2 Solutions Homogeneous mixtures of two or more than two components are called solutions. In the year 1886, Francois Marté a French chemist proposed a law.

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There are three types of solutions:- True solutions:- the solution in which the size of solute particles is less than 10^{-9} m or 1 nm ($< 10^{-9}$ m). Example. solution of sugar in water. Colloidal solution:- the solution in which the size of solute particles is in between 10^{-9} to 10^{-6} m are called colloids. Example. Water in milk.

CBSE Class 12 Chemistry Notes: Surface Chemistry ...

Revision Notes on Solution: A solution is a homogeneous mixture of two (or more) substances, the composition of which may vary between certain limits. A solution consisting of two components is called binary solution. The component which is present in large quantity is called solvent and the component which is small in quantity is called solute.

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