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and its development by Dimitri Mendeleev, elements, how to read individual listings of elements on the table, periods, groups, metals and their physical properties, nonmetals and their physical properties, metalloids, and chemical families.

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9. Describe an unsaturated solution: 10.

Describe a saturated solution: 11.

Describe a supersaturated solution: 12.

What happens when you add more solute to an unsaturated solution? 13.

What happens when you add more solute to a saturated solution?

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For webquest or practice, print a copy of this quiz at the Chemistry: Solutions webquest print page. About this quiz: All the questions on this quiz are based on information that can be found at Chemistry: Solutions. Instructions: To take the quiz, click on the answer. The

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A day at the beach... Tan Without the Sun What makes self-tanning solutions work?

## **Spring Break Chemistry - A WebQuest**

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Webquest Date \_\_\_\_\_ Period \_\_\_\_ Click on the following website to go to your first site ... Diluted solutions have a \_\_\_\_\_ amount of solute. 9. Describe an unsaturated solution: 10. Describe a saturated solution: 11. What happens when you add more solute to an unsaturated solution?

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reactions involve the movement of electrons, leading to the formation and breaking of chemical bonds. Neither matter nor energy is created or destroyed in a chemical reaction---only changed.

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Title: ThermoChemistry (Energy)

WebQuest Author: CCSD Last modified

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Element Project. A WebQuest for High School Chemistry. Designed by.

Charlotte M. Trout mailto:

troutcha@wcboe.k12.md.us.

Introduction | Task | Process | Evaluation

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| Conclusion | Credits. Introduction. By now, you have learned about the structure of the atom and understand something about the location of electrons within the electron cloud around the nucleus.

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