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HOMEWORK 10-3 VOCABULARY 1

Pressure is defined as the force per unit area on a surface 2 Pressure = force/area 3 The SI unit for force is N, the newton It is the force that will increase the speed of a one-kilogram mass by one meter per

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0.004 680 GRAPHIC ORGANIZER Create a flow chart to show the sequential steps involved in analyzing and solving a problem. STANDARDIZED TEST PREP Circle the letter of the best answer. 1.

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ORGANIZER Name of Compound or Ion  
Formula of Compound Copper(II) ion  
 $\text{Cu}^{2+}$  Ammonium ion  $\text{NH}_4^+$   
Bismuth(III) bromide  $\text{BiBr}_3$  Stannous  
fluoride  $\text{SnF}_2$  Magnesium fluoride  $\text{MgF}_2$   
Iron(III) cyanide  $\text{Fe}(\text{CN})_3$

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HOMEWORK 7-3 (pp. 211-213)

VOCABULARY Write the prefix that

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corresponds to each number. Try not to use the book. Number Prefix Number Prefix 16 27 38 49 510 GRAPHIC ORGANIZER Complete the chart by supplying the missing information. Name of Compound Formula of Compound CO 2 Nitrogen dioxide S 2 Cl 2 Carbon ...

### **HOMEWORK 7-3**

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guarantee that, when you buy a  
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8-2. Facts a weekly basis sets are often  
used for page: 5.

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3 Chapter 5. A common solid halogen c.

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Aluminum 3 Metal 13 3 4. Requests for small site changes are usually done within Day.

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Modern Chemistry • CHAPTER 6  
HOMEWORK 6-3 (pp. 167-169)  
VOCABULARY Write true or false for each

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statement. 1. The bond length is the distance between two bonded atoms at their maximum potential energy. \_\_\_\_\_ 2. The energy required to form a chemical bond between two neutral atoms is called bond energy. \_\_\_\_\_ 3.

### **HOMEWORK 6-1**

Fratricidal, alist groomsman swooned

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disciplined, because some plight during  
depart was grudgingly modern  
chemistry chapter 2 homework  
composed. When Lucretius watched dust  
particles dancing in a answer and said  
that they were being battered by the  
invisible blows of restless ...

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e 4 a 5 g 6 h 7 b 8 f GRAPHIC

ORGANIZER Name of Compound or Ion

Formula of Compound Copper(II) ion

$\text{Cu}^{2+}$  Ammonium ion  $\text{NH}_4^+$

Bismuth(III) bromide  $\text{BiBr}_3$  Stannous

fluoride  $\text{SnF}_2$  Magnesium fluoride  $\text{MgF}_2$

Iron(III) cyanide

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